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APPLICATION NO.	FILING DATE	FIRST NAMED INVENTOR	ATTORNEY DOCKET NO.	CONFIRMATION NO.
10/634,964	08/05/2003	Zhendong Liu	02039US	6941
61611	7590 08/18/2006		EXAM	INER
	D HAAS ELECTRONIC	GEORGE, PATRICIA ANN		
CMP HOLDI 451 BELLEV	•		ART UNIT	PAPER NUMBER
NEWARK, DE 19713			1765	
			DATE MAILED: 08/18/200	6

Please find below and/or attached an Office communication concerning this application or proceeding.

	Application No.	Applicant(s)
Office Assistance Occurrence	10/634,964	LIU, ZHENDONG
Office Action Summary	Examiner	Art Unit
	Patricia A. George	1765
The MAILING DATE of this communication a Period for Reply	appears on the cover sheet wi	th the correspondence address
A SHORTENED STATUTORY PERIOD FOR REF WHICHEVER IS LONGER, FROM THE MAILING - Extensions of time may be available under the provisions of 37 CFR after SIX (6) MONTHS from the mailing date of this communication If NO period for reply is specified above, the maximum statutory peri - Failure to reply within the set or extended period for reply will, by sta Any reply received by the Office later than three months after the may earned patent term adjustment. See 37 CFR 1.704(b).	DATE OF THIS COMMUNIC 1.136(a). In no event, however, may a re- tiod will apply and will expire SIX (6) MON tute, cause the application to become AB	CATION. eply be timely filed THS from the mailing date of this communication. BANDONED (35 U.S.C. § 133).
Status		
1) Responsive to communication(s) filed on 20) June 2006.	
2a) ☐ This action is FINAL . 2b) ☑ T	his action is non-final.	
3) Since this application is in condition for allow	wance except for formal matt	ers, prosecution as to the merits is
closed in accordance with the practice unde	er <i>Ex par</i> te Quayle, 1935 C.D). 11, 453 O.G. 213.
Disposition of Claims		
4) ⊠ Claim(s) 1,2 and 4-6 is/are pending in the a 4a) Of the above claim(s) 6-9 is/are without 5) □ Claim(s) is/are allowed. 6) ⊠ Claim(s) 1,2,4,5,6 is/are rejected. 7) □ Claim(s) is/are objected to. 8) □ Claim(s) are subject to restriction and	Irawn from consideration.	
Application Papers		
9) The specification is objected to by the Exam	iner.	
10) The drawing(s) filed on is/are: a) a	accepted or b) objected to	by the Examiner.
Applicant may not request that any objection to t	the drawing(s) be held in abeyar	nce. See 37 CFR 1.85(a).
Replacement drawing sheet(s) including the corr	•	
11)☐ The oath or declaration is objected to by the	Examiner. Note the attached	d Office Action or form PTO-152.
Priority under 35 U.S.C. § 119		
12) Acknowledgment is made of a claim for fore a) All b) Some * c) None of: 1. Certified copies of the priority docume 2. Certified copies of the priority docume 3. Copies of the certified copies of the p application from the International Bur * See the attached detailed Office action for a	ents have been received. ents have been received in A priority documents have been reau (PCT Rule 17.2(a)).	application No received in this National Stage
Attachment(s)		
1) Notice of References Cited (PTO-892) 2) Notice of Draftsperson's Patent Drawing Review (PTO-948) 3) Information Disclosure Statement(s) (PTO-1449 or PTO/SB/Paper No(s)/Mail Date	Paper No(Summary (PTO-413) s)/Mail Date nformal Patent Application (PTO-152)

Application/Control Number: 10/634,964

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DETAILED ACTION

Claim Rejections - 35 USC § 103

The following is a quotation of 35 U.S.C. 103(a) which forms the basis for all obviousness rejections set forth in this Office action:

(a) A patent may not be obtained though the invention is not identically disclosed or described as set forth in section 102 of this title, if the differences between the subject matter sought to be patented and the prior art are such that the subject matter as a whole would have been obvious at the time the invention was made to a person having ordinary skill in the art to which said subject matter pertains. Patentability shall not be negatived by the manner in which the invention was made.

Claims 1, 2, 4, 5, and 6 are rejected under 35 U.S.C. 103(a) as being unpatentable over Sun et al (US Patent No. 6,709,316) in view of Sherber et al. (5,858,813), evidenced by Changzhou Kejia Chemical Co. (Product Detail Poly (Maleic Acid)) and Wikipedia; Maleic Acid (http://www.echinachem.com/kejiachem.co/Product Detail.aspx?Gold Product Profile (http://www.echinachem.com/kejiachem.co/Product Detail.aspx?Gold Product Profile

The reference of Sun et al discloses an aqueous chemical mechanical planarizing composition (col. 9, I. 14). The composition comprises an oxidizer in the form of promoting barrier removal since the composition is the same (col. 6, lines 57-59); an abrasive in the form of (col. 7, lines 9-12); an inhibitor in the form of (for decreasing removal of a metal interconnect) (col. 6, lines 59-65). In addition, Sun et al. clearly discloses ph-adjusting agent/s can adjust the CMP composition to a range of about 2.5 to about 11, which encompassed the claimed less than 4. Sun et al. also clearly explain the adjusting agents can be comprised of bases, inorganic acids, and/or

organic acids (col.6, l.66 to col.7, l.3). Sun teaches a tantalum nitride removal rate of at least eighty percent of copper removal rate (col.3, lines32-34). Sun cites a 1:0:0 removal rate which indicates that 100% of the barrier layer (i.e. TaN) is removed relative to 0% of the copper and 0% of the dielectric.

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In addition Sun teaches a pad pressure of 13.8 kPa. (col. 12, line 1) which is within Sun's range of 1 to 8 psi (6.895 kPa to 55.158 kPa). Sun broadly discloses the use of a chelating agent (col.6, lines 49-54), including the use of carboxyl acids (see claim 10).

Sun fails to disclose the use of water-soluble polymers comprises polymaleic acid (as in claim 1); or palymaleic acid comprises a homo or copolymer (as in claim 2).

Sherber teaches a composition for metal layers and other films during the various stages of multilevel interconnect (i.e. barrier films such as TaN) comprises the conventional use of derivatives from maleic acid, i.e. a water-soluble homopolymer polymaleic acid [evidenced by Wikipedia; Maleic Acid (for solubility) and Changzhou Keija Chemical Co. (maleic acid is homopolymer)], because they posses the ability to complex or associate with dissolving metals and improve the removal rate of metal thin films such as titanium, and the like, during the CMP process (see col. 5, lines 43-58).

It would have been obvious to one of ordinary skill in the art at the time of invention was made, to include a water soluble polymaleic acid, as Sherber, when forming the CMP composition of Sun, because Sherber teaches use of derivatives of the water soluble organic acid, maleic acid, posses the ability to complex with the

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dissolving metals to improve the removal rate of the metal films, a well known process improvement.

As to claim 4, see Sun's teaching on pH above, also see Sherber's teaching that pH can be increased or decreased for a desired range (col.4, lines 41-42).

It would have been obvious to one of ordinary skill in the art at the time the invention was made, to select any desired range of pH, between 2.5 and 11, as Sun, to provide desired results, as Sherber, because overlapping ranges have been held to be obvious.

As to claim 5, Sun teaches an aqueous chemical mechanical planarizing composition comprising 0.05 to 15 wt % abrasive (col. 7, lines 9-12); 0.1 to 10 wt % oxidizing agent (col. 6, lines 57-59); and 0.02 to 1 wt% benzotriazole (col. 6, lines 59-65) which falls within the range of the instant invention. In addition, Sun et al. clearly discloses ph-adjusting agent/s can adjust the CMP composition to a range of about 2.5 to about 11, which encompassed the claimed less than 4. Sun et al. also clearly explain the adjusting agents can be comprised of bases, inorganic acids, and/or organic acids (col.6, l.66 to col.7, l.3). Furthermore, Sun et al. teaches use of phosphoric and nitric acids as a pH-adjusting agents (col.7, I. 3-4 and I.59). Sun et al. teaches a tantalum nitride removal rate of at least eighty percent of copper removal rate (col.3, lines32-34). Sun cites a 1:0:0 removal rate which indicates that 100% of the barrier layer (i.e. TaN) is removed relative to 0% of the copper and 0% of the dielectric

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at a pad pressure of 13.8 kPa. (col. 12, line 1) with in Sun's range of 1 to 8 psi (6.895 kPa to 55.158 kPa). Sun's range is within the claimed range.

Sun fails to disclose the use of water-soluble polymers.

Sherber teaches a composition for metal layers and other films during the various stages of multilevel interconnect (i.e. barrier films such as TaN) comprises the conventional use of derivatives from maleic acid, i.e. a water-soluble homopolymer polymaleic acid [evidenced by Wikipedia; Maleic Acid (for solubility) and Changzhou Kejia Chemical Co. (maleic acid is homopolymer)], because they posses the ability to complex or associate with dissolving metals and improve the removal rate of metal thin films such as titanium, and the like, during the CMP process (see col. 5, lines 43-58).

It would have been obvious to one of ordinary skill in the art at the time of invention was made, to include a water soluble polymaleic acid, as Sherber, when forming the CMP composition of Sun, because Sherber teaches use of derivatives of the water soluble organic acid, maleic acid, posses the ability to complex with the dissolving metals to improve the removal rate of the metal films, a well known process improvement.

As to claims 6, see the discussion above toward claim 2.

Response to Remarks

As to applicants' remarks, in regard to the Declaration of applicant Dr. Liu, please see Response to Declaration, starting on page 5, of the action filed 3/21/2006.

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Applicant's arguments, on page 6, paragraph 1, filed 6/20/2006, with respect to the reference of Yano has been fully considered and is persuasive. The rejection of 3/21/2006 has been withdrawn.

Applicants' continue to argue, on page 5 as in remarks of 10/21/2005, that the reference of Sun et al. does not disclose a second step which operates at a pH below 4, adjusted with an inorganic acid for use in a barrier slurry. Sun et al. clearly discloses ph adjusting agent/s can adjust the CMP composition to a range of about 2.5 to about 11, which encompassed the claimed less than 4. Sun et al. also clearly explain the adjusting agents can be comprised of bases, inorganic acids, and/or organic acids (col.6, 1.66 to col.7, 1.3). That the reference of Sun et al. does not disclose a selective second step is noted but not persuasive because the tantalum removal rate of the solution is properly ascribed to the solution and not an active process step.

Conclusion

The prior art made of record and not relied upon is considered pertinent to applicant's disclosure. Colloidal Dynamics; The CMP Slurry Monitor - Background.

Any inquiry concerning this communication or earlier communications from the examiner should be directed to Patricia A. George whose telephone number is (571) 272-5955. The examiner can normally be reached on weekdays from 7:00am to

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4:30pm. If attempts to reach the examiner by telephone are unsuccessful, the examiner's supervisor, Nadine Norton can be reached on (571) 272-1465. The fax phone number for the organization where this application or proceeding is assigned is 703-872-9306.

Information regarding the status of an application may be obtained from the Patent Application Information Retrieval (PAIR) system. Status information for published applications may be obtained from either Private PAIR or Public PAIR. Status information for unpublished applications is available through Private PAIR only. For more information about the PAIR system, see http://pair-direct.uspto.gov. Should you have questions on access to the Private PAIR system, contact the Electronic

Business Center (EBC) at 866-217-9197 (toll-free).

P/AG/ 08/06 NADINE NORTON EXAMINER.

WORTON EXAMINER.

WORTON EXAMINER.